pH (Hydrogen Ion) Electrometric Method SM 21 st 4500 H ⁺ B Page 1 of 1									
Facility Name:		/ELAP	ID						
Assessor Name:Analyst Name:	Analyst Name:				Inspection Date				
Relevant Aspect of Standards	Method Reference	Υ	N	N/A	Comments				
Records Examined: SOP Number/ Revision/ Date				Analyst:					
Sample ID: Date of Sample Prepa	Date of Sample Preparation:				Date of Analysis:				
Is the meter accurate and reproducible to 0.1 pH unit and equipped with a temperature-compensation adjustment?	4500-H ⁺ B 2.a								
Is the electrode storage solution in accordance with manufacturer's instructions?	4500-H ⁺ B 4.a								
If a nonsealed electrode is used, is it filled with the correct electrolyte to the proper level?	4500-H ⁺ B 2.b								
Is the meter calibrated at least once daily (on days of use) following manufacturer's instructions?	4500-H ⁺ B 4.a								
Are buffer solutions within the listed expiration, or have they been prepared within the last four weeks?	4500-H ⁺ B 3.a								
Do the buffer solutions appear to be free of growths or contamination?	4500-H ⁺ B 3.a								
Is the cap or sleeve covering the access hole on the reference electrode removed when measuring pH?	Manufacturer's Instructions? (See method section 2.b)								
Is the sample stirred gently at a constant speed during measurement?	4500-H ⁺ B 4.b								
Does the meter hold a steady reading after reaching equilibrium?	4500-H ⁺ B 4.b								
Are the electrodes rinsed and blotted dry with a soft tissue before each reading? (Disregard if a portion of the next sample to be analyzed is used as a rinse.)	4500-H⁺B 4.a								
Is the temperature of buffer solutions and samples recorded when determining pH? (Only if using a meter with a temperature dial.)	4500-H⁺B 4.a								
Following calibration, is a buffer analyzed as a check sample with an acceptable range of ± 0.1 pH unit from the true value of the buffer?	4500-H⁺B 4.a								

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